**Today** 

- I. Last Time
- II. Wrap up Identical Particle Distribution Functions III. Degenerate Fermi Gas
- I. Spencer gave a guest lecture on Gibbs factors and the grand partition function

 $e^{-(E(s)-\mu N(s))/kT}$ .

The grand partition function is sum over these Gibbs factors

$$
\mathscr{Z} = \sum_{s} e^{-(E(s) - \mu N(s))/kT}.
$$

The setup for these is a system in contact with a constant temperature reservoir that can also exchange particles.



I. We found the grand partition function for fermions

$$
\mathcal{Z} = 1 + e^{-(\epsilon - \mu)/kT}.
$$

From this we found the "occupancy", that is, the average number of particles in the state under consideration. We found

$$
\overline{n}_{FD} = \frac{1}{e^{(\epsilon - \mu)/kT} + 1}.
$$



I. Let's turn to the grand partition function for bosons  $= 1 + e^{-(\epsilon - \mu)/kT} + (e^{-(\epsilon - \mu)/kT})^2 + \cdots$  $\mathscr{L} = \sum e^{-n(\epsilon - \mu)/kT} = 1 + e^{-(\epsilon - \mu)/kT} + e^{-2(\epsilon - \mu)/kT} + \cdots$ *n* 2  $+ \cdots$ = 1  $1 - e^{-(\epsilon - \mu)/kT}$ 

Notice that this sum only converges if the term being raised to a power has a magnitude less than 1. Here this amounts to the requirement that  $\epsilon$  be greater than  $\mu$ !

Once again we compute occupancy via  
\n
$$
\overline{n} = \sum_{n} nP(n) = OP(0) + 1P(1) + \cdots.
$$

Let's introduce the shorthand  $x \equiv (\epsilon - \mu)/kT$ ...

II. We have:  $\mathscr{Z} =$ Once again we compute occupancy via 1  $1 - e^{-(\epsilon - \mu)/kT}$ 

$$
\bar{n} = \sum_{n} nP(n) = OP(0) + 1P(1) + \cdots
$$

Let's introduce the shorthand  $x \equiv (\epsilon - \mu)/kT$ ...

$$
\overline{n}_{BE} = \sum_{n} n \frac{e^{-nx}}{\mathcal{L}} = \frac{-1}{\mathcal{L}} \sum_{n} \frac{\partial}{\partial x} e^{-nx} = -\frac{1}{\mathcal{L}} \frac{\partial \mathcal{L}}{\partial x} = \frac{1}{e^{(\epsilon - \mu)/kT} - 1}
$$

If we had stuck with Boltzmann statistics we would have found

$$
P(s) = \frac{1}{Z_1}e^{-\epsilon/kT}, \overline{n}_{Boltzmann} = NP(s) = \frac{N}{Z_1}e^{-\epsilon/kT} = e^{\mu/kT}e^{-\epsilon/kT} = e^{-(\epsilon-\mu)/kT}.
$$

On the homework you're showing that  $\mu = -kT \ln(Z_1/N)$ 

## II. This is nicely summarized graphically.



Figure 7.7. Comparison of the Fermi-Dirac, Bose-Einstein, and Boltzmann distributions, all for the same value of  $\mu$ . When  $(\epsilon - \mu)/kT \gg 1$ , the three distributions become equal.

## III. Degenerate Fermi Gases

To fix a context let's imagine the electrons in a metal as our fermions.

$$
v_Q = \left(\frac{h}{\sqrt{2\pi m kT}}\right)^3
$$
 = (4.3 nm)<sup>3</sup> (Room temperature gas of electrons)

Compare an electron per atom, which gives  $(0.1 \text{ nm})^3$ , which much smaller.

This shows that we are well away from the dilute limit and we should think about the Fermi-Dirac statistics. In that case, why not think of this as a zero temperature limit. We'll genuinely justify this after the fact.

## III. Degenerate Fermi Gases

This shows that we are well away from the dilute limit and we should think about the Fermi-Dirac statistics. In that case, why not think of this as a zero temperature limit. We'll genuinely justify this after the fact. In the zero temperature limit the chemical potential becomes the deciding factor as to whether a state is occupied or not and we call it the "Fermi energy" of the system:

$$
\epsilon_F \equiv \mu(T=0).
$$



III. Fermi energy:

$$
\epsilon_F \equiv \mu(T=0).
$$

Let's again fix our attention on a cubical box of side length L (we're think of it as a chunk of metal). Recall that free particles in such a box have

 $\lambda_n = \frac{2L}{n}$ , and  $p_n = \frac{hc}{\lambda} = \frac{hc}{\lambda}$ , but this time let's take into account the 2*L n*  $p_n =$ *h λn* = *hn* 2*L*

3D nature of the box

 $p_x = \frac{\Delta p_x}{2I}, p_y = \frac{y}{2I}, p_z = \frac{\Delta p_z}{2I}$ . The corresponding energies  $\epsilon = \frac{P}{2} = \frac{n}{R} = \frac{n}{R_x^2} (n_x^2 + n_y^2 + n_z^2).$  $hn_{\chi}$  $\frac{x}{2L}$ ,  $p_y =$  $hn_{y}$  $\frac{\partial}{\partial L}$ ,  $p_z =$  $hn_z$ 2*L*  $|\overrightarrow{p}|^2$ 2*m* =  $h^2$  $\frac{n}{8mL^2}(n_x^2 + n_y^2 + n_z^2)$