

QM 4/13/15 (MATT)



1. GENERAL SCHRÖDINGER EQN. FOR SPHERICALLY SYMMETRIC POTENTIALS

• 
$$-\frac{\hbar^2}{2m} \nabla^2 \psi(\vec{r}) + V(r) \psi(\vec{r}) = E \psi(\vec{r})$$

• DO SEPARATION OF VARIABLES,  $\psi(\vec{r}) = R(r) \cdot Y(\theta, \phi)$   
 FIND SPHERICAL HARMONICS SOLUTIONS,  
 WITH  $L^2 = \hbar^2 \cdot l(l+1)$

• GIVES RADIAL EQN.

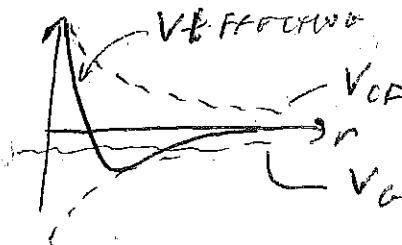
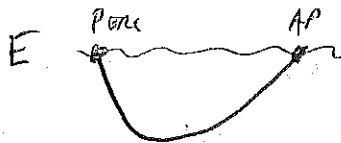
$$-\frac{\hbar^2}{2m} \frac{1}{r^2} \frac{d}{dr} \left( r^2 \frac{dR}{dr} \right) + V(r) R(r) + \frac{\hbar^2 l(l+1)}{2mr^2} R = ER$$

2. CONCEPTUALLY,  $\frac{\hbar^2 l(l+1)}{2mr^2}$  IS "CENTRIFUGAL BARRIER".

FOR LARGE  $l$ , IT IS HARDER FOR A PARTICLE TO HAVE A SMALL RADIUS.

•  $V(r) + \frac{\hbar^2 l(l+1)}{2mr^2}$  IS "EFFECTIVE POTENTIAL".

e.g. WITH PLANETS,



FOR A GIVEN ENERGY, PERIHELION AND APHELION FOR  $\dot{r} = 0$   
 ( $K_{RADIAL} = 0$ , ONLY  $K_{ROTATIONAL}$ )

3. ODE IS EASIER TO SOLVE USING  $u \equiv r \cdot R$ .

• SO  $R = \frac{1}{r} u$ ,  $R' = \frac{1}{r} u' - \frac{1}{r^2} u$

$r^2 R' = (r u' - u)$

$(r^2 R')' = u' + r u'' - u' = r u''$

$$-\frac{\hbar^2}{2m} \frac{u''}{r} + \left[ V(r) + \frac{\hbar^2 l(l+1)}{2mr^2} \right] \frac{u}{r} = E \frac{u}{r} \quad \text{SIMILAR}$$

4. MOST OF QM IS SOLVING THIS EQUATION FOR DIFFERENT CHOICES OF  $V(r)$ , ANALOGOUS TO WHAT YOU DID IN 1-DIMENSION

- 3-D WELL } EMILY'S
- 3-D HARMONIC OSCILLATOR } SPROJ
- 3-D H-ATOM

5. 3-D SPHERICAL WELL  $V(r) = \begin{cases} 0 & r < a \\ \infty & r > a \end{cases}$

$$\frac{d^2 u}{dr^2} = \left[ \frac{l(l+1)}{r^2} - k^2 \right] u$$

• For  $l=0$ ,  $u = \sin(kr)$

• For  $l > 0$ ,  $u$  MORE COMPLICATED BUT SIMILAR (BESSEL)

AS ALWAYS IN QM, QUANTIZATION COMES FROM APPLYING THE BOUNDARY CONDITIONS, IN THIS CASE THAT  $u(a) = 0$ .

$u(ka) = 0$  MEANS  $ka$  IS A ZERO OF THE BESSEL F.W. (FIGURE 4.2)

THIS DETERMINES  $k$ , AND  $\boxed{\frac{\hbar^2 k^2}{2m} = E}$

GET QUANTIZED ENERGIES.

6. YOU CAN EXTEND TO FINITE WELL, SOLVE IT NUMERICALLY (EMILY AGAIN)

7. H-ATOM : FUNDAMENTAL TEST CASE FOR QM

8. ON WEDNESDAY, MAYA WILL DISCUSS  
THE OBSERVED LINE SPECTRA

$$\frac{1}{\lambda} = R \left[ \frac{1}{n_1^2} - \frac{1}{n_2^2} \right] \quad R = 0.01097 \text{ nm}^{-1} \quad (\text{RYDBERG})$$

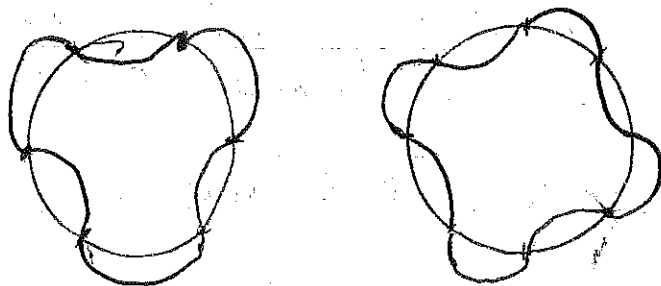
9. WHAT NEEDS TO BE EXPLAINED BY A THEORY:

- ① WHY DOESN'T H-ATOM ALWAYS RADIATE?
- ② WHAT IS THE PROCESS OF EMISSION OR ABSORPTION OF LIGHT?
- ③ WHY THESE PARTICULAR  $\lambda$ 'S AND ONLY THESE?

10. 3 LEVELS OF 'ANSWERS', SOME SIMILARITIES!

11. BOHR: ①: THERE ARE STABLE ORBITS WHEN  $L = n\hbar$   
(TALK THROUGH DERIVATION ON WANDS) ② ELECTRON TRANSITION  $E_f = \Delta E_e = |E_n - E_{n'}|$   
③ DERIVATION GIVES RYDBERG FORMULA  
FROM  $\frac{1}{\lambda} = \frac{E_f}{hc}$

12. DEBROGLIE: ① STABLE ORBITS ARE DUE TO STANDING WAVES.  
②, ③ SAME ANSWER AS BOHR.



13. QM: (1) WHEN A SYSTEM IS IN AN ENERGY EIGENSTATE, IT DOES NOT LOSE ENERGY  
 (2) SAME ANSWER AS BOHR AND DE BROGLIE  
 (3) SOLVE SCHRÖDINGER EQN. FOR EIGENSTATES

$$-\frac{\hbar^2}{2m} \frac{d^2 u}{dr^2} + \left[ -\frac{\hbar^2 k^2}{2m} + \frac{\hbar^2 l(l+1)}{2mr^2} \right] u = Eu.$$

14. FINDING POLYNOMIAL SOLUTIONS (START [3])

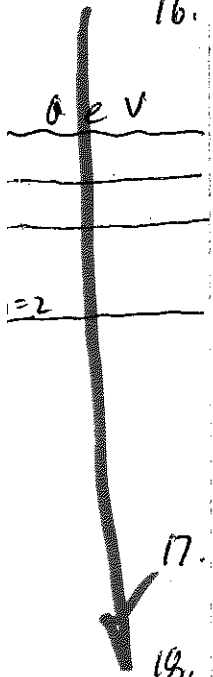
15. LAGUERRE POLYNOMIALS IN CONTEXT OF EARLIER SOLUTION SETS

- SIN/COS IN 1-D SQ. WELL
- HERMITE POLYS IN 1-D HARMONIC OSCILLATOR
- BESSEL FNS IN 3-D SPHERICAL WELL

16. DIDN'T GET TO THIS

$E_n$	$n, l, m$	$d$
-0.85 eV	4s, 4p, 4d, 4f	(16)
-1.51 eV	3s, 3p, 3d	(9)
-3.40 eV	2s, 2p	(4)
-13.6 eV	1s	(1)

$n = 1, 2, 3, \dots$   
 $l = 0, 1, 2, \dots, n-1$   
 $m = -l, -(l-1), \dots, 0, \dots, (l-1), l$



$$\psi_{n\ell m}(r, \theta, \phi) = \left[ \left( \frac{2}{na} \right)^3 \frac{(n-\ell-1)!}{2^n (n+\ell)!} \right]^{1/2} e^{-r/na} \cdot \left( \frac{2r}{na} \right)^\ell \cdot L_{n-\ell-1}^{2\ell+1} \left( \frac{2r}{na} \right) \cdot Y_\ell^m(\theta, \phi)$$

17. QUALITATIVE DISCUSSION OF WAVE FNS.

18. NOTICE:  $E(n)$  ONLY, NOT  $l$  OR  $m$ .

- SPHERICAL SYMMETRY  $\Rightarrow$   $m$ -DEGENERACY
- MORE SUBTLE SYMMETRY  $\Rightarrow$   $l$ -DEGENERACY.

$l=0, l=0$

HYDROGEN ATOM - POLYNOMIAL SOLN

1. SPH. SYM. POT  $\Rightarrow u(r) \equiv r \cdot R(r)$   
 SPH. HARM  $\Rightarrow V_L = \frac{\hbar^2 l(l+1)}{2mr^2}$  CONTRIBUTION TO SPH. POT.

H-ATOM  $-\frac{\hbar^2}{2m} \frac{d^2 u}{dr^2} + \left[ -ke \frac{1}{r} + \frac{\hbar^2 l(l+1)}{2mr^2} \right] u = Eu \quad \underline{E < 0}$

2.  $\frac{\hbar^2}{2m}$  HAS UNITS  $E \cdot L^2$ , SO DEFINE A LENGTH  $1/k$  CHARACTERISTIC  
 $k^2 \equiv \frac{-E}{\frac{\hbar^2}{2m}} \Rightarrow k \equiv \frac{\sqrt{-2mE}}{\hbar} \quad \left[ = \frac{1}{a_0} \text{ (IT WILL TURN OUT)} \right]$

$$\therefore \frac{1}{k^2} \frac{d^2 u}{dr^2} = \left[ 1 - \frac{2mke}{\hbar^2 k} \left( \frac{1}{kr} \right) + \frac{l(l+1)}{(kr)^2} \right] u$$

3. DEFINE DIM. LESS RADIUS VARIABLE:  $\rho \equiv kr$   
 FOR P.E. TERM:  $\rho_0 \equiv \frac{2mke}{\hbar^2 k}$
- $$d^2 u'' = \left[ 1 - \frac{\rho_0}{\rho} + \frac{l(l+1)}{\rho^2} \right] u$$

4. COULD JUST TRY  $u = \sum a_n \rho^n$ , BUT NOT WORKING (IT TURNS OUT)

$\rho \gg 1/\rho_0 \quad u'' \approx u$   
 $u = A e^{-\rho} + B e^{+\rho} \quad BC \Rightarrow B = 0$

$\rho \rightarrow 0, \quad u'' \approx \frac{l(l+1)}{\rho^2} u$   
 $u = \rho^\alpha \Rightarrow \alpha(\alpha+1) = l(l+1)$   
 $\therefore \alpha = -l \text{ or } \alpha = l+1$   
 $BC \Rightarrow \text{ONLY } (l+1)$

$\therefore u(\rho) = \rho^{l+1} e^{-\rho} \cdot v(\rho)$   
 $\hookrightarrow$  SOLVE FOR THIS  $v(\rho)$

$$5. \quad \rho v'' + 2(l+1-\rho)v' + [\rho_0 - 2(l+1)]v = 0$$

$$\text{TRY } v(\rho) = \sum_0^{\infty} a_j \rho^j$$

$$v' = \sum (j+1) a_{j+1} \rho^j \quad \rho v' = \sum j a_j \rho^j$$

$$\rho v'' = \sum j(j+1) a_{j+1} \rho^j$$

$$\rho^j: \quad j(j+1) a_{j+1} + 2(l+1)(j+1) a_{j+1} - 2j a_j + [\rho_0 - 2(l+1)] a_j = 0$$

$$a_{j+1} = \left[ \frac{2(j+l+1) - \rho_0}{(j+1)(j+2l+2)} \right] a_j$$

RECURSION REL.

$$6. \text{ If } j \gg l, \quad a_{j+1} \sim \frac{2j}{j^2} \Rightarrow v \sim e^{-2\rho}$$

$$\text{BUT THEN } u = A \rho^{2l+1} e^{-\rho} \text{ DIVERGES}$$

DIDN'T WRITE HAS TO TERMINATE

$$\text{FOR SOME } j, \quad 2(j+l+1) - \rho_0 = 0.$$

GET

$$\text{DPPING } n = j_{\max} + l + 1, \quad \rho_0 = 2n \quad \boxed{l \geq n-1}$$

TO  
THIS

$$E_n = -\frac{\hbar^2 k^2}{2m} = -\frac{1}{2} (m c^2) \left( \frac{k e}{\hbar c} \right)^2 \cdot \frac{1}{n^2} \\ = -13.6 \text{ eV} \cdot \frac{1}{n^2}$$

$$E = \frac{1}{2} \rho e \quad \therefore -\frac{1}{2} \frac{k e}{\rho} \Rightarrow -13.6 \text{ eV} \Rightarrow \rho_0 = \rho_0$$

2. DISCUSS DIFF

PASSED OUT  
AND SKIPPED  
THROUGH IT

### Bohr Atom "Derivation"

#### Classical Circular Orbits

$$\frac{k_e}{r^2} = \frac{mv^2}{r} \quad \left( \text{with } k_e \equiv \frac{e^2}{4\pi\epsilon_0} \right)$$

Centripetal Force for a Circular Orbit

$$\frac{k_e}{r} = mv^2$$

$$E = \left( \frac{1}{2} mv^2 - \frac{k_e}{r} \right) = \left( -\frac{1}{2} mv^2 \right) = \left( -\frac{1}{2} \frac{k_e}{r} \right)$$

Total Energy = Kinetic + Potential

$$|\vec{L}| = |\vec{r} \times \vec{p}| = r m v$$

Angular Momentum for Circular Orbit

$$E = \left( -\frac{L^2}{2mr^2} \right) = \left( -\frac{1}{2} \frac{k_e}{r} \right)$$

#### Bohr Quantization

$$L_n = n \hbar$$

Bohr's Quantization Condition

$$\frac{k_e}{r_n} = \frac{(n \hbar)^2}{m r_n^2}$$

$$r_n = \left( \frac{\hbar^2}{k_e m} \right) n^2 = a n^2$$

Quantized Orbital Radii

$$E_n = \left( -\frac{1}{2} \frac{k_e}{r_n} \right) = -E_1 \frac{1}{n^2}$$

Quantized Energy Levels

#### Numerical Values

Useful Constants:	$\hbar c = 197.37 \text{ eV} \cdot \text{nm}$	$mc^2 = 511,000 \text{ eV}$	$\frac{k_e}{\hbar c} = \frac{1}{137.04}$
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Bohr Radius: 
$$a \equiv \left( \frac{\hbar^2}{k_e m} \right) = (\hbar c) \left( \frac{\hbar c}{k_e} \right) \left( \frac{1}{mc^2} \right) = 0.0529 \text{ nm}$$

Lowest Energy: 
$$E_1 \equiv \frac{1}{2} \frac{k_e}{a} = \frac{1}{2} \frac{k_e^2 m}{\hbar^2} = \frac{1}{2} \left( \frac{k_e}{\hbar c} \right)^2 (mc^2) = 13.6 \text{ eV}$$

