<u>Today</u>

- I. Last Time
- II. Completing Our Discussion of the Radial Equation III. The 'Old' Quantum Theory of Hydrogen
- I. Last time
- * Separation of variables to arrive at the radial and angular Schrodinger equations in 3D.
- * Found the eigenstates of L^2 and L_z , which turned out to be the spherical harmonics $Y_{\ell}^m(\theta, \phi)$.
- * One intuition behind these functions is that they provide a basis for all functions on the sphere. We can also think of them as describing the quantum states of definite magnitude of angular momentum and definite z-component of ang. mom. These are made up of $e^{im\phi}$ and solutions to Legendre's equation.

I. Derivation of angular momentum eigenfunctions.

Let's focus on (1) at first $\frac{1}{\Phi} \frac{d^2 \Phi}{d\phi^2} = -m^2 \quad (1)$

It has solutions

 $\Phi(\phi) = e^{im\phi}.$

Boundary conditions tell us the allowed values for m, in the present case they are

 $\Phi(\phi + 2\pi) = \Phi(\phi).$ Let's impose this $e^{im(\phi + 2\pi)} = e^{im\phi} \implies e^{im2\pi} = 1,$ The last condition is satisfied when $m = 0, \pm 1, \pm 2, ..., \pm \ell$. I. Derivation of angular momentum eigenfunctions.

Let's focus on (2) at first

$$\left\{\frac{1}{\Theta}\left[s\theta\frac{d}{d\theta}\left(s\theta\frac{d\Theta}{d\theta}\right)\right] + \ell(\ell+1)s^2\theta\right\} = m^2 \quad (2)$$

It has solutions

 $\Theta(\theta) = AP_{\ell}^{m}(\cos\theta),$

where P_{ℓ}^{m} is the associated Legendre polynomial (or function),

$$P_{\ell}^{m}(x) = (1 - x^{2})^{|m|/2} \left(\frac{d}{dx}\right)^{|m|} P_{\ell}(x),$$

which in turn is defined in terms of the Legendre polynomial

$$P_{\ell}(x) = \frac{1}{2^{\ell}\ell!} \left(\frac{d}{dx}\right)^{\ell} (x^2 - 1)^{\ell}.$$

This last formula is called the Rodrigues formula.

I. Derivation of angular momentum eigenfunctions.

This all amounts to having found the "spherical harmonics"

$$Y_{\ell}^{m}(\theta,\phi) = \Phi(\phi)\Theta(\theta) = \sqrt{\frac{(2\ell+1)(\ell-|m|)!}{4\pi(\ell+|m|)!}}e^{im\phi}P_{\ell}^{m}(\cos\theta).$$

These are also orthogonal (why?). So,

 $\int_{0}^{2\pi} \int_{\theta=0}^{\pi} \left[Y_{\ell}^{m}(\theta,\phi) \right]^{*} \left[Y_{\ell'}^{m'}(\theta,\phi) \right] \sin \theta d\theta d\phi = \delta_{\ell\ell'} \delta_{mm'}.$ (Beware the

Condon-Shortley phase. Also the Geodesy convention, which is a real basis for spherical harmonics.)

Our first example of a basis for functions were sin and cos in the context of Fourier analysis. The intuition for spherical harmonics is exactly the same—they provide a basis of functions for the sphere.

I. The Radial Equation

In these spherical coordinates

$$\nabla^2 = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 s \theta} \frac{\partial}{\partial \theta} \left(s \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{r^2 s^2 \theta} \frac{\partial^2}{\partial \phi^2}$$

Notice that part of this (the last two terms) is exactly $-\frac{1}{\hbar^2 r^2}L^2$.

Once again we're tempted to try and use separation of variables $\psi(r, \theta, \phi) = R(r)Y(\theta, \phi)$. Then

$$-\frac{\hbar^2}{2m}\nabla^2(RY) + V(RY) = E(RY)$$

$$\implies -\frac{\hbar^2}{2mr^2}Y\frac{\partial}{\partial r}\left(r^2\frac{dR}{dr}\right) + \frac{\hbar^2}{\hbar^22mr^2}RL^2Y + VRY = ERY$$

I. The Radial Equation

$$-\frac{\hbar^2}{2mr^2}Y\frac{\partial}{\partial r}\left(r^2\frac{dR}{dr}\right) + \frac{\hbar^2}{\hbar^22mr^2}RL^2Y + VRY = ERY$$

Dividing by *RY* and multiplying by $-2mr^2/\hbar^2$:
$$\left\{\frac{1}{R}\frac{d}{dr}\left(r^2\frac{dR}{dr}\right) - \frac{2mr^2}{\hbar^2}\left[V(r) - E\right]\right\} + \frac{1}{Y}\left\{\frac{1}{s\theta}\frac{\partial}{\partial\theta}\left(s\theta\frac{\partial Y}{\partial\theta}\right) + \frac{1}{s^2\theta}\frac{\partial^2 Y}{\partial\phi^2}\right\} = 0$$

II. Completing Our Discussion of the Radial Equation

Notice that the second operator is ang. mom. squared roughly, $-\frac{1}{\hbar^2 Y}L^2Y = -\ell(\ell+1),$

So the solutions to the angular equation are the spherical harmonics. This means that

$$\frac{1}{R}\frac{d}{dr}\left(r^2\frac{dR}{dr}\right) - \frac{2mr^2}{\hbar^2}[V(r) - E] = \ell(\ell+1)$$

II. Completing Our Discussion of the Radial Equation This means that

$$\frac{1}{R}\frac{d}{dr}\left(r^2\frac{dR}{dr}\right) - \frac{2mr^2}{\hbar^2}[V(r) - E] = \ell(\ell+1);$$

let's study this equation. Let's change variables to

 $u(r) = rR(r) \quad \text{or} \quad R(r) = u/r. \text{ Let's compute derivatives}$ $\frac{dR}{dr} = \frac{r\frac{du}{dr} - u}{r^2},$ $\frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) = r^2 \frac{d^2 u}{dr^2}.$

Then our differential equation in terms of *u* becomes

$$-\frac{\hbar^2}{2m}\frac{d^2u}{dr^2} + \left[V(r) + \frac{\hbar^2}{2m}\frac{\ell(\ell+1)}{r^2}\right]u = Eu$$

Or

 $-\frac{\hbar^2}{2m}\frac{d^2u}{dr^2} + V_{\text{eff}}u = Eu, \text{ where } V_{\text{eff}}(r) = V(r) + \frac{\hbar^2}{2m}\frac{\ell(\ell+1)}{r^2}$

II. Completing Our Discussion of the Radial Equation

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Let's consider a particular potential, just to get a feel for what this looks like. The hydrogen atom potential is

$$V(r) = -\frac{1}{4\pi\epsilon_0} \frac{e^2}{r} = -\frac{k_e e^2}{r},$$

then the effective potential is
$$V_{\text{eff}} = -\frac{k_e e^2}{r} + \frac{\hbar^2}{2m} \frac{\ell(\ell+1)}{r^2}$$



III. The 'Old' Quantum Theory of Hydrogen (The Bohr Model)

Classical Circular Orbit

centripetal force for a circular orbit

This gives

 $\frac{mv^2}{r} = \frac{k_e e^2}{r^2}$

$$\frac{k_e e^2}{r} = mv^2$$

The total energy in the system (KE+PE),

$$E = \left(\frac{1}{2}mv^2 - \frac{k_e e^2}{r}\right) = -\frac{1}{2}mv^2 = -\frac{1}{2}\frac{k_e e^2}{r}$$

The angular momentum of this circular orbit is $|\vec{L}| = |\vec{r} \times \vec{p}| = rmv$, And

$$E = \left(-\frac{L^2}{2mr^2}\right) = -\frac{1}{2}\frac{k_e e^2}{r}.$$

