

# Today

- I. Last Time
- II. Finish Power Series Method for Hydrogen
- III. Saiqi's Guest Lecture on Spin

## I. Last time

\*Intro'd a power series solution to try and solve the Hydrogen atom radial wave function: part of that was  $v(\rho) = \sum_{j=0}^{\infty} c_j \rho^j$ . (Recall  $v$

followed on from  $u = rR(r)$ .)

\*The power series  $v$  came from stripping off the asymptotic behavior of  $u = \rho^{\ell+1} e^{-\rho} v(\rho)$ , where  $\rho = kr$ .

# I. Our Result for the Radial Equation

$$-\frac{\hbar^2}{2m} \frac{d^2 u}{dr^2} + V_{\text{eff}} u = E u, \quad \text{where} \quad V_{\text{eff}}(r) = V(r) + \frac{\hbar^2}{2m} \frac{\ell(\ell + 1)}{r^2}$$

Let's consider a particular potential, just to get a feel for what this looks like. The hydrogen atom potential is

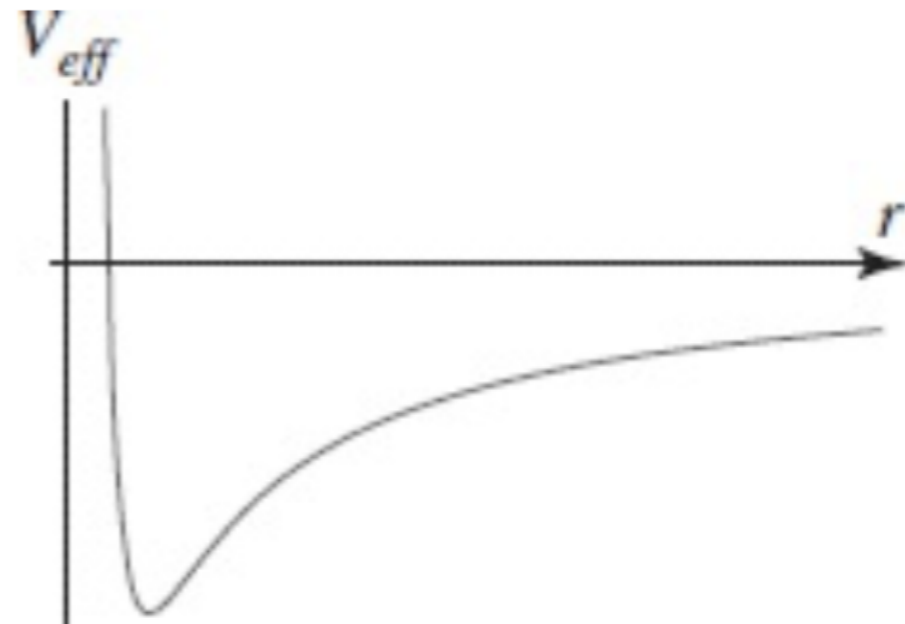
$$V(r) = -\frac{1}{4\pi\epsilon_0} \frac{e^2}{r} = -\frac{k_e e^2}{r},$$

then the effective potential is

$$V_{\text{eff}} = -\frac{k_e e^2}{r} + \frac{\hbar^2}{2m} \frac{\ell(\ell + 1)}{r^2}.$$

Here little  $u$ ,

$$u(r) = rR(r).$$



# I. Series Method for Hydrogen atom

Let's introduce a wave number for bound states

$$k = \frac{\sqrt{-2m_e E}}{\hbar}.$$

Plugging this in we find

$$\frac{1}{k^2} \frac{d^2 u}{dr^2} = \left[ 1 - \frac{m_e e^2}{2\pi\epsilon_0 \hbar^2 k} \frac{1}{(kr)} + \frac{\ell(\ell+1)}{(kr)^2} \right] u,$$

Let's introduce nice variables

$$\rho = kr \quad \text{and} \quad \rho_0 = \frac{m_e e^2}{2\pi\epsilon_0 \hbar^2 k}.$$

Then we have the clean form

$$\frac{d^2 u}{d\rho^2} = \left[ 1 - \frac{\rho_0}{\rho} + \frac{\ell(\ell+1)}{\rho^2} \right] u.$$

Let's look for the asymptotic behavior of this equation, let's take up the case  $\rho \rightarrow \infty \dots$



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Let's look for the asymptotic behavior of this equation, let's take up the case  $\rho \rightarrow \infty$  where

$$\frac{d^2u}{d\rho^2} = u,$$

and the solutions are just

$$u(\rho) = Ae^{-\rho} + Be^{\rho}.$$

There's no way that the second term is normalizable, so we only keep

$$u(\rho) \sim Ae^{-\rho}.$$

The other asymptotic regime is  $\rho \rightarrow 0$ , where

$$\frac{d^2u}{d\rho^2} = \frac{\ell(\ell + 1)}{\rho^2}u, \quad \text{with solution } u(\rho) = C\rho^{\ell+1} + D\rho^{-\ell} \text{ (check it).}$$



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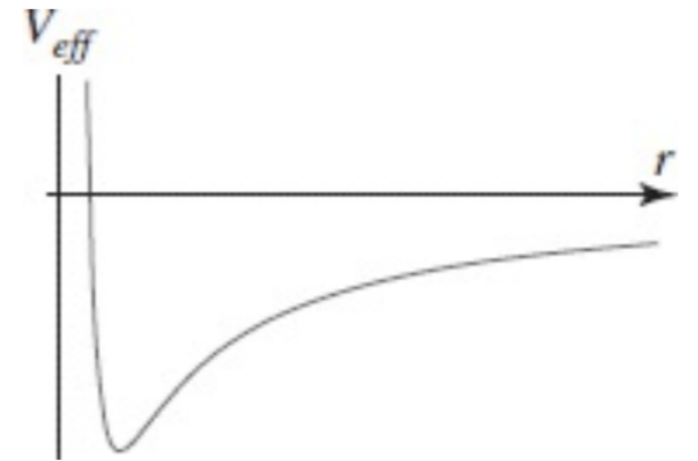
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normalizable one is  $u \sim C\rho^{\ell+1}$ . Then our full ansatz for the power series method is going to be

$u(\rho) = \rho^{\ell+1}e^{-\rho}v(\rho)$ . Putting this in the resulting differential equation is

$$\rho \frac{d^2v}{d\rho^2} + 2(\ell+1-\rho) \frac{dv}{d\rho} + [\rho_0 - 2(\ell+1)]v = 0$$



# I. Series Method for Hydrogen atom

Starting with

$$\rho \frac{d^2 v}{d\rho^2} + 2(\ell + 1 - \rho) \frac{dv}{d\rho} + [\rho_0 - 2(\ell + 1)]v = 0,$$

we now guess a solution of the form

$$v(\rho) = \sum_{j=0}^{\infty} c_j \rho^j.$$

Now we grind

$$\frac{dv}{d\rho} = \sum_{j=0}^{\infty} j c_j \rho^{j-1} = \sum_{j=0}^{\infty} (j+1) c_{j+1} \rho^j$$

And for the second derivative

$$\frac{d^2 v}{d\rho^2} = \sum_{j=0}^{\infty} j(j+1) c_{j+1} \rho^{j-1}$$



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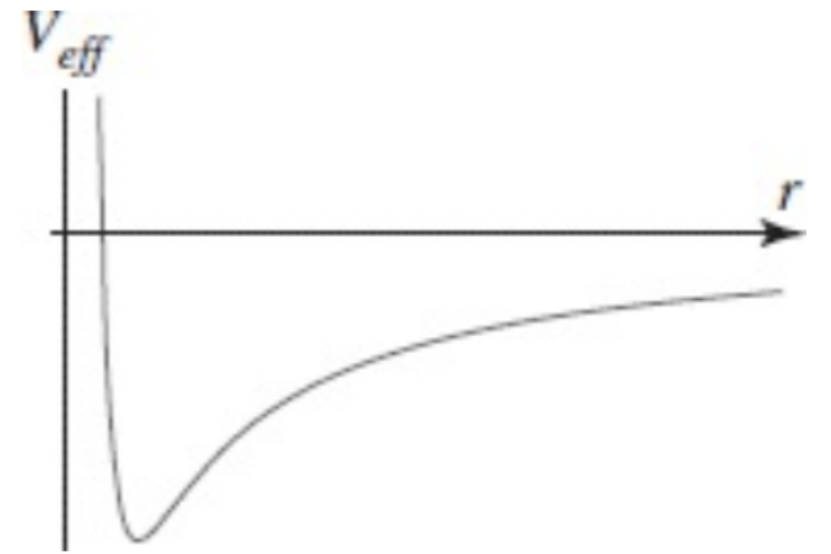
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And for the second derivative

$$\frac{d^2 v}{d\rho^2} = \sum_{j=0}^{\infty} j(j+1) c_{j+1} \rho^{j-1}.$$

$$\sum_{j=0}^{\infty} \left[ j(j+1) c_{j+1} \rho^j + 2(\ell + 1) j(j+1) c_{j+1} \rho^j - 2j c_j \rho^j + (\rho_0 - 2(\ell + 1)) c_j \rho^j \right] = 0$$



## I. Series Method for Hydrogen atom

$$\sum_{j=0}^{\infty} \left[ j(j+1)c_{j+1}\rho^j + 2(\ell+1)j(j+1)c_{j+1}\rho^j - 2jc_j\rho^j + (\rho_0 - 2(\ell+1))c_j\rho^j \right] = 0$$

Setting the coefficients of every term equal to zero we learn that

$$\left[ j(j+1)c_{j+1} + 2(\ell+1)j(j+1)c_{j+1} - 2jc_j + (\rho_0 - 2(\ell+1))c_j \right] = 0.$$

This gives the recursion relation:

$$c_{j+1} = \frac{2(j+\ell+1) - \rho_0}{(j+1)(j+2\ell+2)} c_j.$$

Let's imagine for a moment that this recursion never ends, then for large  $j$ , we can approximate it in a simpler form:

$$c_{j+1} = \frac{2j}{j(j+1)} c_j = \frac{2}{j+1} c_j.$$

This new recursion and it has an approximate solution

$$c_j \approx \frac{2^j}{j!} c_0, \text{ which implies } v(\rho) \approx c_0 \sum_{j=0}^{\infty} \frac{2^j}{j!} \rho^j = c_0 e^{2\rho}; \text{ nota bene(!!!)...}$$



## II. Series Method for Hydrogen atom

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...this is a bad solution! It doesn't meet our boundary conditions, in particular, we can't normalize it. This means that we can't allow the recursion to continue to large  $j$ . So, instead we force it to end:

$$c_{j+1} = \frac{2(j + \ell + 1) - \rho_0}{(j + 1)(j + 2\ell + 2)} c_j \text{ implies that the numerator should vanish}$$

for some value of  $j$ , which we will call  $N$ . Then

$2(N + \ell) - \rho_0 = 0$ , then with  $n \equiv N + \ell$ , we get  $\rho_0 = 2n$ . We now have that

$$E = -\frac{\hbar^2 k^2}{2m} = -\frac{m_e e^4}{8\pi^2 \epsilon_0^2 \hbar^2 \rho_0^2} \implies E_n = -\left[ \frac{m_e}{2\hbar^2} \left( \frac{e^2}{4\pi\epsilon_0} \right)^2 \right] \frac{1}{n^2} = \frac{E_1}{n^2}$$

## II. Series Method for Hydrogen atom

Then the total wave function is

$$\psi_{n\ell m}(r, \theta, \phi) = R_{n\ell}(r)Y_{\ell}^m(\theta, \phi),$$

Where

$$R_{n\ell}(r) = \frac{1}{r}\rho^{\ell+1}e^{-\rho}v(\rho),$$

Where  $v(\rho)$  is determined by

$$c_{j+1} = \frac{2(j + \ell + 1 - n)}{(j + 1)(j + 2\ell + 2)}c_j.$$

$$\text{Also } \rho \equiv \frac{r}{an}.$$